organic papers

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Rong Wan,* Feng Wu, Jian-Ling Guan, Feng Han and Jin-Tang Wang

Department of Applied Chemistry, College of Science, Nanjing University of Technology, No. 5 Xinmofan Road, Nanjing, Nanjing 210009, People's Republic of China

Correspondence e-mail: rwan01@jlonline.com

Key indicators

Single-crystal X-ray study T = 298 KMean $\sigma(\text{C}-\text{C}) = 0.005 \text{ Å}$ R factor = 0.053 wR factor = 0.146 Data-to-parameter ratio = 14.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-(4-Fluorophenyl)-3-[5-(4-methoxyphenyl)-1,3,4-thiadiazol-2-yl]thiazolidin-4-one

In the title compound, $C_{18}H_{14}FN_3O_2S_2$, the thiazolidine ring adopts an envelope conformation. There are intramolecular $C-H\cdots S$ and intermolecular $C-H\cdots O$ interactions, forming chains along the *b* axis.

Comment

Thiadiazole derivatives containing the thiazolidinone unit are of great interest because of their chemical and pharmaceutical properties. Some derivatives have fungicidal activities and exhibit some herbicidal activities (Chen *et al.*, 2000; Kidwai *et al.*, 2000; Vicentini *et al.*, 1998), and some show insecticidal activities (Arun *et al.*, 1999; Wasfy *et al.*, 1996).



We are focusing our synthetic and structural studies on thiadiazole derivatives and we have published recently the structure of 3-[5-(4-fluorophenyl)-1,3,4-thiadiazol-2-yl]-2-phenylthiazolidin-4-one (Wan *et al.*, 2006). We report the crystal structure of a close analog, (I), in which the 4-fluorophenyl substituent is replaced by 4-methoxyphenyl, and 2-phenyl is replaced by 4-fluorophenyl. In (I), the thiazolidine ring adopts an envelope conformation with atom C7 at the tip of the flap (Fig. 1 and Table 1). There are intramolecular C–H···S and intermolecular C–H···O interactions (Table 2), forming chains along the *b* axis (Fig. 2).





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Experimental

(4-Fluorobenzylidene)[5-(4-methoxyphenyl)-1,3,4-thiadiazol-2-yl]amine (5 mmol) and 3-mercaptopropionic acid (5 mmol) were dissolved in toluene (50 ml). The resulting water was removed by distillation over a period of 5 h. The reaction mixture was left to cool to room temperature and filtered; the solid was recrystallized from acetone to give (I) (yield 81.3%, m.p. 520–524 K). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of an acetone solution.

Z = 8

 $D_x = 1.462 \text{ Mg m}^{-3}$ Mo K\alpha radiation $\mu = 0.33 \text{ mm}^{-1}$ T = 298 K Block, colorless 0.40 \times 0.40 \times 0.30 mm

3449 independent reflections

 $R_{\rm int} = 0.089$

 $\theta_{\rm max} = 26.0^{\circ}$

3 standard reflections

every 200 reflections

intensity decay: none

2562 reflections with $I > 2\sigma(I)$

Crystal data

$C_{18}H_{14}FN_3O_2S_2$
$M_r = 387.44$
Monoclinic, C2/c
a = 17.750 (4) Å
b = 6.3720 (13) Å
c = 31.761 (6) Å
$\beta = 101.55 \ (3)^{\circ}$
$V = 3519.5 (12) \text{ Å}^3$

Data collection

Enraf–Nonius CAD-4 diffractometer $\omega/2\theta$ scans Absorption correction: ψ scan (North *et al.*, 1968) $T_{\min} = 0.879, T_{\max} = 0.907$ 3562 measured reflections

Refinement

Table 1

Selected torsion angles (°).

C9-N1-C7-S1	9.3 (3)	C7-N1-C9-C8	-5.7 (4)
C8-S1-C7-N1	-8.0(2)	S1-C8-C9-N1	-1.2(3)
C7-S1-C8-C9	5.5 (2)		

Table 2

Hydrogen-bond	geometry	(Å.	°)
i i yui ogen-oonu	geometry	(д,)

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$\begin{array}{c} \hline \text{C7}-\text{H7}A\cdots\text{O1}^{\text{i}}\\ \text{C13}-\text{H13}A\cdots\text{S2} \end{array}$	0.98	2.49	3.299 (4)	140
	0.93	2.77	3.163 (3)	107

Symmetry code: (i) x, y + 1, z.

All H atoms were placed geometrically, with C—H = 0.93–0.98 Å, and included in the refinement in the riding-model approximation with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ or $1.5U_{\rm eq}({\rm methyl~C})$.



Figure 2

Part of the crystal structure of (I). Dashed lines indicate $C{-}H{\cdots}O$ interactions.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Siemens, 1996); software used to prepare material for publication: *SHELXL97*.

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